Statistical Mechanics (4)

<u>Def 4.1</u>

Consider a gas inside a piston. Define dQ to be the thermal energy transferred from the surrounding to the system. $\dots \dots (4.1.1)$

Define dU, the change of internal energy of the gas to be

$$dU = dQ - PdV \qquad \dots \dots (4.1.2)$$

Define molar heat capacity of the gas at constant volume be

$$C_{\nu} = \frac{1}{\mu} \left(\frac{\partial Q}{\partial T}\right)_{\nu} \qquad \dots \dots \dots (4.1.3)$$

where μ is the no. of mole of the gas. Define the molar heat capacity of the gas at constant pressure to be

$$C_{P} = \frac{1}{\mu} \left(\frac{\partial Q}{\partial T}\right)_{P} \qquad \dots \dots (4.1.4)$$

Define $\gamma = \frac{C_{P}}{C_{V}} \qquad \dots \dots (4.1.5)$

<u>Thm 4.2</u>

Suppose a gas in a piston obey ideal gas law (i.e. $PV = \mu RT = NkT$) and that $(\frac{\partial U}{\partial V})_T = 0$ and that γ is a constant (independent of any variable), then it can be proven that if the gas undergo adiabatic process (i.e. dQ=0), then *P*, *V* must obey the following rule

$$PV^{\gamma} = \text{constant}$$
(4.2.1)

<u>Thm 4.3</u>

It can be verified from experiment that for all kind of gas in piston,

$$\left(\frac{\partial U}{\partial V}\right)_T = 0 \tag{4.3.1}$$

This law is called the Joule's Law.

<u>Def 4.4</u>

If a gas is piston obey ideal gas law, it can be proven that

$$C_P - C_V = R \tag{4.4.1}$$

Def 4.5

A process undergo by a gas in a piston is reversible iff

- I. The process is very slow such that at each time in the process, the system is in equilibrium (settled down)
- II. The work done by the system must equal to PdV (i.e. no energy dissipation due to friction, etc.)

<u>Thm 4.6</u>

Suppose function $f = f(x_1, ..., x_n)$ is to be maximized under the constraint

$$\begin{cases} g_1(x_1,...,x_n) = 0 \\ \vdots \\ g_p(x_1,...,x_n) = 0 \\ \dots \dots \dots (4.6.1) \end{cases}$$

Then the problem can be solved by solving the simultaneous equations:

which involve n+p unknown: $x_1, ..., x_n$; $\lambda_1, ..., \lambda_p$ and n+p equations.

<u>Thm 4.7</u>

 $\sum_{i=1}^{i} n_i$ identical distinguishable particle is to distribute along the energies, such that the no. of cell with energy ε_i is g_i and the no. of particle with energy ε_i is n_i , then the no. of way is given by

$$W(n_1, n_2, ...) = \frac{(N(n_1, n_2, ...))!}{n_1! n_2! \cdots} * g_1^{n_1} g_2^{n_2} \cdots$$
(4.7.1)

where
$$N(n_1, n_2, ...) = \sum_i n_i$$
(4.7.2)

Define the entropy $S(n_1, n_2, ...) = k \ln W(n_1, n_2, ...)$ (4.7.3)

Let $n_i >> 1$ for all *i*, then using Stirling formula $(\ln n! = n \ln n - n \text{ for } n >> 1)$, we can prove that

$$S(n_1, n_2, ...) = k[N(n_1, ...) \ln N(n_1, ...) - \sum_i n_i \ln n_i + \sum_i n_i \ln g_i] \qquad(4.7.4)$$

If *S* is to maximize under the constraint

$$g_1(n_1, n_2, ...) = \{\sum_i n_i\} - N_0 = 0$$

$$g_2(n_1, n_2, ...) = \{\sum_i n_i\} - E_0 = 0$$

.....(4.7.5) and (4.7.6)

then according to Thm 4.6,

$$\frac{1}{k}\frac{\partial S}{\partial n_j} - \alpha \frac{\partial g_1}{\partial n_j} - \beta \frac{\partial g_2}{\partial n_j} = 0$$
 for $j = 1, 2, \dots$ (4.7.7)

which will lead to that

$$n_j = N_0 g_j e^{-\alpha} e^{-\beta \varepsilon_j}$$
 for $j = 1, 2,$ (4.7.8)

Suppose after *S* is maximized under the two constraint (4.7.5), (4.7.6), suddenly the total internal energy is changed by dU. Let the change of no. of particle with energy ε_i is dn_i .

Since
$$U(n_1, n_2, ...) = \sum_i n_i \varepsilon_i$$
, then

$$dU = \sum_{i} \frac{\partial U}{\partial n_{i}} dn_{i} = \sum_{i} \varepsilon_{i} dn_{i} \qquad \dots \dots (4.7.9)$$

From (4.7.7), we have

$$\frac{1}{k}\frac{\partial S}{\partial n_j} - \alpha - \beta \varepsilon_j = 0 \qquad \text{for } j = 1, 2, \dots \qquad \Rightarrow \sum_j (\frac{1}{k}\frac{\partial S}{\partial n_j} - \alpha - \beta \varepsilon_j) dn_j = 0$$

Let the total no. of particle is not changed:

$$\Rightarrow \frac{1}{k}dS = \beta \sum_{j} \varepsilon_{j} dn_{j} = \beta dU = \beta dQ$$
 (Assume no volume change)

By comparing with the formula $dS = \frac{dQ}{T}$, we have $\beta k = \frac{1}{T}$ or

$$\beta = \frac{1}{kT} \tag{4.7.10}$$

<u>Thm 4.8</u>

 $\sum_{i=1}^{i} n_i$ identical indistinguishable particles is to distribute along the energies, such that the no. of cell with energy ε_i is g_i and the no. of particles with energy ε_i is n_i , then the no. of way is given by(4.8.0)

$$W(n_1, n_2, ...) = \prod_i \frac{(n_i + g_i - 1)!}{n_i!(g_i - 1)!}$$
(4.8.1)

Let $n_i >> 1$ for all I, and use Stirling formula, we can prove that

$$S(n_1, n_2, ...) = k \sum_i [(n_i + g_i - 1) \ln(n_i + g_i - 1) - n_i \ln n_i - (g_i - 1) - \ln(g_i - 1)!]$$

$$g_1(n_1, n_2, ...) = \{\sum_i n_i \varepsilon_i\} - E_o = 0$$
(4.8.2)

If S is to maximize under the constraint $\sum_{i=1}^{N_1, N_2, \dots, N_i} \sum_{i=1}^{N_i, N_i}$

$$\frac{1}{k}\frac{\partial S}{\partial n_j} - \beta \frac{\partial g_1}{\partial n_j} = 0$$
 for $j=1,2,...$ (4.8.3)

which lead to $n_j = \frac{g_j - 1}{e^{\beta \varepsilon_j} - 1} \approx \frac{g_j}{e^{\beta \varepsilon_j} - 1}$ (if $g_i >>1$)(4.8.4)

<u>Thm 4.9</u>

Assume in a cubic box of side L, and set the periodic boundary condition to photon:

Let $\rho^2 = n_1^2 + n_2^2 + n_3^2$ (4.9.2) Then the no. of photon between ρ and $\rho + d\rho$ is $2(4\pi\rho^2)d\rho$ (4.9.3) where the factor 2 comes from that photon have two direction of polarization. Substitute

(4.9.1), (4.9.2) into (4.9.3), we have the no. of photon between v and v+dv (v is

frequency) is $\frac{2(4\pi(\frac{L}{c})^3 v^2)dv}{c}$ (4.9.4) and the total energy of photons between v and v+dv is therefore by substitution into (4.8.4):

$$dE = \frac{hv(2)4\pi (\frac{L}{c})^3 v^2 dv}{e^{\frac{hv}{kT}} - 1}$$

and so energy volume density of photon between v and v+dv

$$\frac{dE}{L^3} = \frac{8\pi h v^3 dv}{c^3 (e^{\frac{hv}{kT}} - 1)}$$
(Planck Radiation Formula)(4.9.5)

By special mathematics technique (Appendix A of "Statistical Physics", by F. Mandl), it can be evaluate that

According to the corpuscle picture of photon, rate of energy passing through a certain area should be linearly related with the energy volume density, therefore, we can infer the wall of the box, which is a blackbody and have the same temperature with the photons radiates as

$$R \propto T^4 = e\sigma T^4$$
 (Stefan-Boltzmann Law)(4.9.7)

where R is the energy per unit area per unit time, $\sigma = \frac{ac}{4}$ is the Stefan constant, *e* is the emissivity. For perfect reflector, *e*=0, and for blackbody, *e*=1.

<u>Thm 4.11</u>

 $\sum_{i}^{i} n_{i}$ identical indistinguishable particles is to distribute along the energies, such that the no. of cell with energy ε_{i} is g_{i} and the no. of particle with energy ε_{i} is n_{i} , and that every cell can only accommodate 1 particle, then the no. of way is given by $\dots (4.11.0)$

$$W(n_1, n_2, ...) = \prod_i \frac{g_i!}{n_i!(g_i - n_i)!}$$
(4.11.1)

Let gi is large enough, such that both g_i , n_i and g_i - n_i are large enough to apply the Stirling formula

$$\begin{cases} \ln g_{i}! \approx g_{i} \ln g_{i} - g_{i} \\ \ln n_{i}! \approx n_{i} \ln n_{i} - n_{i} \\ \ln (g_{i} - n_{i})! \approx (g_{i} - n_{i}) \ln (g_{i} - n_{i}) - (g_{i} - n_{i}) \end{cases} \qquad (4.11.2)$$

Then the entropy can be given by

$$S(n_1,...) = k \sum_i [g_i \ln g_i - g_i - n_i \ln n_i + n_i - (g_i - n_i) \ln(g_i - n_i) + (g_i - n_i)]$$
(4.11.3)

If S is to maximize under the constraint

$$g_1(n_1, n_2, ...) = \{\sum_i n_i\} - N_0 = 0$$

$$g_2(n_1, n_2, ...) = \{\sum_i n_i\} - E_0 = 0$$

.....(4.11.4) and (4.11.5)

Then according to Thm 4.6,

$$\frac{1}{k}\frac{\partial S}{\partial n_j} - \alpha \frac{\partial g_1}{\partial n_j} - \beta \frac{\partial g_2}{\partial n_j} = 0$$
 for $j = 1, 2, \dots$ (4.11.6)

which will lead to that

<u>Thm 4.12</u>

Since $\lambda v = c \Rightarrow v = c/\lambda \Rightarrow dv = -\frac{c}{\lambda^2} d\lambda$. So from (4.9.5), we have

$$\frac{dE}{L^3} = \frac{8\pi h(\frac{c}{\lambda^3})\frac{c}{\lambda^2}d\lambda}{c^3(e^{hc/\lambda kT}-1)} = \frac{8\pi hcd\lambda}{\lambda^5(e^{hc/\lambda kT}-1)} \qquad \dots \dots (4.12.1)$$

So the energy volume density of photon between $(\lambda, \lambda + d\lambda)$ is given by (4.12.1). To find the value of λ where $\frac{1}{L^3} \frac{dE}{d\lambda}$ is maximum, we have

$$\frac{d}{d\lambda}(\frac{1}{L^3}\frac{dE}{d\lambda}) = 0 \qquad \dots \dots (4.12.2)$$

From (4.12.2), we have λ_{max} , the value of λ where $\frac{1}{L^3} \frac{dE}{d\lambda}$ is maximum is given by

 $\lambda_{\text{max}}T = 2.898 \times 10^{-3}$ (Wien's displacement Law)(4.12.3)

<u>Thm 4.13</u>

When $\frac{1}{L^3} \frac{dE}{d\lambda}$ in (4.9.5) is plotted against v, we have

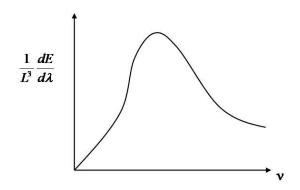


Figure 4.13.1

<u>Thm 4.14</u>

From formula (4.7.3), we know that $S = S(n_1, n_2,...)$. If for every moment, *S* is to be maximized under the constraint (4.7.5) & (4.7.6), we have in (4.7.8) $n_i = n_i(N, E)$, where *N*, *E* is the total no. of particle and total energy respectively. Then $S = S(n_1(N, E), n_2(N, E),...)$. Define the chemical potential

$$\mu = -T(\frac{\partial S}{\partial N})_E \qquad \dots \dots (4.14.1)$$

From (4.7.7), we have

$$\frac{1}{k}\frac{\partial S}{\partial n_{i}}(N,E) - \alpha - \beta \varepsilon_{i} = 0$$

$$\Rightarrow \frac{1}{k}\sum_{i}\frac{\partial S}{\partial n_{i}}(N,E)\frac{\partial n_{i}}{\partial N} = \sum_{i}(\alpha + \beta \varepsilon_{i})\frac{\partial n_{i}}{\partial N}$$
L.H.S. $= \frac{\partial S}{\partial N}$
Consider the functions
$$f_{1}(N,E) = \sum_{i}n_{i}(N,E) = N \Rightarrow \frac{\partial f_{i}}{\partial N} = \sum_{i}\frac{\partial n_{i}}{\partial N} = 1$$

$$f_{2}(N,E) = \sum_{i}n_{i}(N,E)\varepsilon_{i} = E \Rightarrow \frac{\partial f_{i}}{\partial E} = 0$$

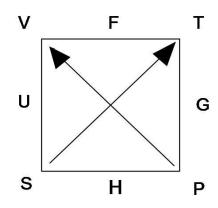
$$\therefore \text{ R.H.S.} = \alpha$$

<u>Thm 4.15</u>

Suppose two variables out of four variables: T, V, S, P are independent variables. Suppose, there exist a U, such that U can be written as the function of any two independent variables out of T, V, S, P and that if we take S, V as the two independent variable, we will have

$$dU = TdS - PdV \qquad \dots \dots (4.15.1)$$

Then we can prove the "Maxwell relations":



$$(\frac{\partial S}{\partial V})_{T} = (\frac{\partial P}{\partial T})_{V} \qquad \dots \dots (4.15.2a) \qquad (\frac{\partial S}{\partial P})_{T} = -(\frac{\partial V}{\partial T})_{P} \qquad \dots \dots (4.15.2b)$$
$$(\frac{\partial V}{\partial S})_{P} = (\frac{\partial T}{\partial P})_{S} \qquad \dots \dots (4.15.2c) \qquad (\frac{\partial T}{\partial V})_{S} = -(\frac{\partial P}{\partial S})_{V} \qquad \dots \dots (4.15.2d)$$

Proof:

Firstly, define the function H (enthalpy), F (Helmhotz's free energy) and G (Gibb's free energy) by

$$H=U+PV$$
......(4.15.3) $F=U-TS$(4.15.4) $G=U-TS+PV$(4.15.5)

Then take the total differentiation dH, dF and dG, and make use of (4.15.1), we will get

dH = TdS + VdP	(4.15.6a)
dF = -SdT - PdV	(4.15.6b)
dG = -SdT + VdP	(4.15.6c)

Let us repeat (4.15.1) here

$$dU = TdS - PdV \qquad \dots \dots (4.15.6d)$$

Then from (4.15.6a) to (4.15.6d), we will be able to prove the Maxwell's relations (4.15.2a) to (4.15.2d) easily.

<u>Thm 4.16</u>

Suppose $\sum_{i}^{i} n_{i}$ identical distinguishable particle is to distribute along a series of cells, such that the *i*th cell has n_i particles. Suppose a particle in the *i*th cell will possess a energy ε_i , then the no. of way is given by

$$W(n_1, n_2, ...) = \frac{(N(n_1, n_2, ...))!}{n_1! n_2! \cdots}$$
(4.16.1)

where $N(n_1, n_2, ...) = \sum_i n_i$(4.16.2)

Define the entropy $S(n_1, n_2, ...) = k \ln W(n_1, n_2, ...)$(4.16.3)

Let $n_i >> 1$ for all *i*, then using Stirling formula $(\ln n! = n \ln n - n \text{ for } n >> 1)$, we can prove that

$$S(n_1, n_2, ...) = k[N(n_1, ...) \ln N(n_1, ...) - \sum_i n_i \ln n_i]$$
(4.16.4)

If *S* is to maximize under the constraint

$$g_1(n_1, n_2, ...) = \{\sum_i n_i\} - N_0 = 0$$

$$g_2(n_1, n_2, ...) = \{\sum_i n_i\} - E_0 = 0$$

.....(4.16.5) and (4.16.6)

then according to Thm 4.6,

$$\frac{1}{k}\frac{\partial S}{\partial n_j} - \alpha \frac{\partial g_1}{\partial n_j} - \beta \frac{\partial g_2}{\partial n_j} = 0$$
 for $j = 1, 2, \dots$ (4.16.7)

which will lead to that

$$n_j = N_0 e^{-\alpha} e^{-\beta \varepsilon_j}$$
 for $j = 1, 2,$ (4.7.8)

<u>Thm 4.17</u>

From (4.15.4), we have F=U-TS. Let us assume every thing in Thm 4.16 still valid. Then

$$F = \sum_{i} n_{i} \varepsilon_{i} - Tk[N \ln N - \sum_{i} n_{i} \ln n_{i}]$$

$$= \sum_{i} n_{i} \varepsilon_{i} - Tk[N \ln N - \sum_{i} n_{i} \ln(Ne^{-\alpha}e^{-\beta\varepsilon_{i}})]$$

$$= \sum_{i} n_{i} \varepsilon_{i} - Tk[N \ln N - \sum_{i} n_{i} (\ln N - \alpha - \beta\varepsilon_{i})]$$

$$= \sum_{i} n_{i} \varepsilon_{i} - Tk[N \ln N - N \ln N + N\alpha + \beta\sum_{i} n_{i} \varepsilon_{i}]$$

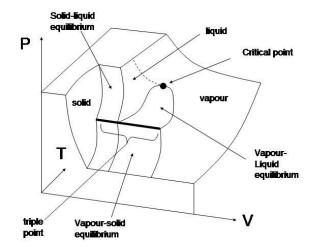
$$= -TkN\alpha$$

As $\sum_{i} N e^{-\alpha} e^{-\beta \varepsilon_i} = N \Longrightarrow e^{-\alpha} (\sum_{i} e^{-\beta \varepsilon_i}) = 1$. Define the partition function

$$Z = \sum_{i} e^{-\beta \varepsilon_{i}} \qquad \dots \dots (4.17.1)$$
$$\Rightarrow e^{\alpha} = Z \Rightarrow \alpha = \ln Z$$
$$\Rightarrow F = -TkN \ln Z \qquad \dots \dots (4.17.2)$$

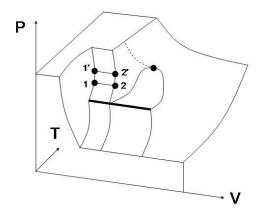
<u>Thm 4.18</u>

A typical PVT system



Critical point
$$(\frac{\partial P}{\partial V})_T = (\frac{\partial}{\partial V}(\frac{\partial P}{\partial V})_T)_T = 0$$

<u>Thm 4.19</u>



Let G_1 , G_2 be the Gibb's free energy at point 1 & 2 respectively. When the system moves along the dashes line from 1 to 2, *P* & *T* keep constant. So the change in Gibb's energy from (4.15.5) is

dG = dU - TdS + PdV = dQ - TdS	(4.19.0a)
$\therefore G_1 = G_2 \therefore \text{Similarly} \ G_{1'} = G_{2'} \Rightarrow$	$G_{1'} - G_1 = G_{2'} - G_2$ (4.19.0b)

From (4.15.6c),

$$G_{1'} - G_1 = -S_1(T_{1'} - T_1) + V_1(P_{1'} - P_1) \qquad \dots \dots (4.19.0c)$$

$$G_{2'} - G_2 = -S_2(T_{2'} - T_2) + V_2(P_{2'} - P_2) \qquad \dots \dots (4.19.0d)$$

Where S_1 , V_1 , S_2 , V_2 are entropy and volume at point 1 & 2 respectively.

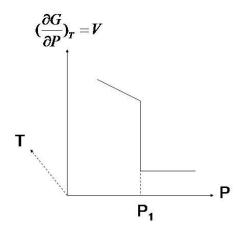
$$\Rightarrow -S_1 dT + V_1 dP = -S_2 dT + V_2 dP$$

$$\Rightarrow (V_2 - V_1) dP = (S_2 - S_1) dT$$

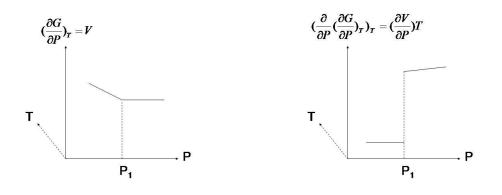
$$\Rightarrow \frac{dP}{dT} = \frac{S_2 - S_1}{V_2 - V_1}$$
 (Clausius-Clapeyron Equation)(4.19.1)

<u>Thm 4.20</u>

In Thm 4.19, $(\frac{\partial G}{\partial P})_T = V$ is discontinuous at a pressure (namely P_1):



Suppose now $(\frac{\partial G}{\partial P})_T = V$ is continuous at P_1 , but $(\frac{\partial}{\partial P}(\frac{\partial G}{\partial P})_T)_T = (\frac{\partial V}{\partial P})T$ is discontinuous at P_1 , i.e.



Then

$$\lim_{P \to P_1^-} d(V) = \lim_{P \to P_1^-} \left(\frac{\partial V}{\partial T}\right)_P dT + \lim_{P \to P_1^-} \left(\frac{\partial V}{\partial P}\right)_T dP \qquad \dots \dots (4.20.0a)$$
$$\lim_{P \to P_1^+} d(V) = \lim_{P \to P_1^+} \left(\frac{\partial V}{\partial T}\right)_P dT + \lim_{P \to P_1^+} \left(\frac{\partial V}{\partial P}\right)_T dP \qquad \dots \dots (4.20.0b)$$

As $\lim_{P \to P_1^-} d(V) = \lim_{P \to P_1^+} d(V)$, we have

$$(\lim_{P \to P_1^-} (\frac{\partial V}{\partial T})_P - \lim_{P \to P_1^+} (\frac{\partial V}{\partial T})_P) dT = (\lim_{P \to P_1^-} (\frac{\partial V}{\partial P})_T - \lim_{P \to P_1^+} (\frac{\partial V}{\partial P})_T) dP$$

$$\frac{dP}{dT} = -\frac{\lim_{P \to P_1^-} (\frac{\partial V}{\partial T})_P - \lim_{P \to P_1^+} (\frac{\partial V}{\partial T})_P}{\lim_{P \to P_1^-} (\frac{\partial V}{\partial P})_T - \lim_{P \to P_1^+} (\frac{\partial V}{\partial P})_T} \qquad \dots \dots (4.20.1)$$

 $(\frac{\partial G}{\partial T})_{P} = -S \text{ is continuous at } T_{1} \text{ but } (\frac{\partial}{\partial T}(\frac{\partial G}{\partial T})_{P})_{P} = -(\frac{\partial S}{\partial T})_{P} \text{ is discontinuous at } T_{1}, \text{ then similar to the above argument,}$

$$\frac{dP}{dT} = \frac{\lim_{T \to T_1^-} (\frac{\partial S}{\partial T})_P - \lim_{T \to T_1^+} (\frac{\partial S}{\partial T})_P}{\lim_{T \to T_1^-} (\frac{\partial V}{\partial T})_P - \lim_{T \to T_1^+} (\frac{\partial V}{\partial T})_P} \qquad \dots \dots (4.20.2)$$

<u>Thm 4.21</u>

Consider the W given in (4.8.1). Suppose for all $i, g_i >> n_i$, then

$$W = \prod_{i} \frac{(g_{i} + n_{i} - 1)(g_{i} + n_{i} - 2)\cdots(g_{i})}{n_{i}!} \approx \prod_{i} \frac{(g_{i})^{n_{i}}}{n_{i}!} \qquad \dots \dots (4.21.1)$$

Consider the *W* given in (4.11.1). Suppose for all *i*, $g_i \gg n_i$, then

$$W = \prod_{i} \frac{(g_{i})!(g_{i}-1)!\cdots(g_{i}-n_{i}+1)!}{n_{i}!} \approx \prod_{i} \frac{(g_{i})^{n_{i}}}{n_{i}!} \qquad \dots \dots (4.21.2)$$

: For Bose-Einstein & Fermi-Dirac System, if $g_i >> n_i$ for all i,(4.21.3) we can use treatment similar to the Maxwell Boltzmann System as in Thm. 4.7.

Thm 4.22

Assume we are talking about the system described in Thm 4.16,

$$-\frac{\partial}{\partial\beta}\ln Z = -\frac{1}{Z}\frac{\partial Z}{\partial\beta} = -\frac{1}{Z}\frac{\partial}{\partial\beta}(\sum_{i}e^{-\beta\varepsilon_{i}}) \quad (\text{From (4.17.1)}) = -\frac{1}{Z}\sum_{i}-\varepsilon_{i}e^{-\beta\varepsilon_{i}} = \frac{1}{Z}\sum_{i}\varepsilon_{i}e^{-\beta\varepsilon_{i}}$$

$$U = \sum_{i}n_{i}\varepsilon_{i} = \sum_{i}Ne^{-\alpha}e^{-\beta\varepsilon_{i}}\varepsilon_{i} \qquad \sum_{i}Ne^{-\alpha}e^{-\beta\varepsilon_{i}} = N$$
By definition
$$\Rightarrow e^{-\alpha}Z = 1 \Rightarrow U = N\frac{1}{Z}\sum_{i}e^{-\beta\varepsilon_{i}}\varepsilon_{i} = -N\frac{\partial}{\partial\beta}\ln Z \qquad \dots (4.22.1)$$

<u>Thm 4.23</u>

Assume we are still talking about the system described in Thm 4.16. Suppose each cell is

formed by a point in the phase space $\bigotimes_{i=1}^{n_{ph}} {\xi_i : \xi_i \in \Re}$ such that the cell at ${\xi_i}_{i=1}^{n_{ph}}$ possess an energy

$$\varepsilon = \sum_{i=1}^{n_{ph}} A_i \xi_i^2$$
(4.23.1)

Then the partition function will be given by (4.17.1)

$$Z = \sum_{j} e^{-\beta(\sum_{i=1}^{n_{ph}} A_i \xi_{i,j}^2)}$$

Let in the phase space $\overset{n_{ph}}{\underset{i=1}{\otimes}} \{\xi_{i,1} : \xi_{i,1} \in (\xi_i, \xi_i + \Delta \xi_i)\}$, the no. of state is given by $g(\xi_1, \dots, \xi_{n_{ph}}) \Delta \xi_1 \cdots \Delta \xi_{n_{ph}}$, then

$$Z = \int_{-\infty}^{+\infty} \cdots \int_{-\infty}^{+\infty} e^{-\beta(\sum_{i=1}^{n_{ph}} A_i \xi_i^2)} g(\xi_1, ..., \xi_{n_{ph}}) d\xi_1 \cdots d\xi_{n_{ph}}$$

Suppose $g(\xi_1,...,\xi_{n_{ph}})$ is a constant, independent of $\xi_1,...,\xi_{n_{ph}}$, then

$$Z = g \prod_{i=1}^{n_{ph}} \int_{-\infty}^{+\infty} e^{-\beta A_i \xi_i^2} d\xi_i = g \prod_{i=1}^{n_{ph}} \left(\frac{\pi}{\beta A_i}\right)^{1/2} = g \beta^{-n_{ph}/2} \prod_{i=1}^{n_{ph}} \left(\frac{\pi}{A_i}\right)^{1/2}$$
$$\Rightarrow \frac{\partial Z}{\partial \beta} = -\frac{n_{ph}}{2} \frac{1}{\beta} Z$$

From (4.22.1),

$$U = -N\frac{1}{Z}\frac{\partial Z}{\partial \beta} = -\frac{N}{Z}(-\frac{n_{ph}}{2\beta})Z = \frac{1}{2}Nn_{ph}kT$$
.....(4.23.2)